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## Phosphorus, Sulfur, and Silicon and the Related Elements

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### *P-Functionalized Phosphaalkenes and Iminophosphines: New Results*

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## P-FUNCTIONALIZED PHOSPHAALKENES AND IMINOPHOSPHINES: NEW RESULTS

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**Abstract** Synthesis, structures, spectroscopic data, and novel reactions of P-[bis(dialkylamino)methyleneamino]- and P-[bis(dialkylamino)methylenephosphino]-substituted phosphalkenes, iminophosphines and phosphonium ions are discussed.

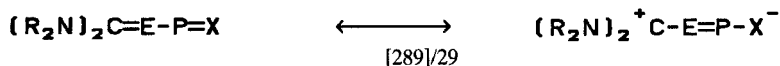
### INTRODUCTION

In recent years we and other authors have reported the synthesis of a number of stable, highly reactive two-coordinate phosphorus derivatives including those with the P-Si, P-Hlg, P-P and P-O bonds.<sup>1</sup> These species were shown to be of considerable interest as important precursors for a broad spectrum of products which can be obtained by nucleophilic or electrophilic substitution without affecting the P=C, P=N or P=P bonds. We have since extended our researches to reaction chemistry of the functionalized phosphonium ions and dienic like systems of the type  $(R_2N)_2C=E-P=X$  (E = N, P).<sup>2-5</sup>

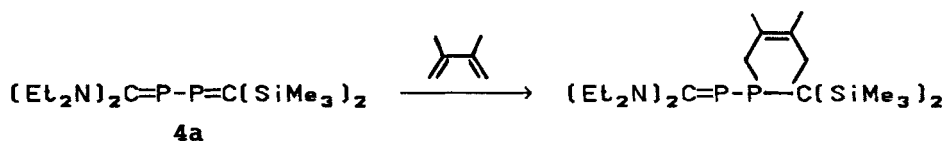
### RESULTS AND DISCUSSION

#### Dienic like systems including dicoordinated phosphorus

Novel stable 2-aza-3-phospha- and 2,3-diphosphabutadienes have been prepared via condensation of the compounds  $(R_2N)_2C=ESiMe_3$  (E = N, P) with  $ClP=X$  (X = CTms<sub>2</sub>, Nar). The structures and properties of these dienic like compounds are of special interest because they can be considered as potentially conjugated systems:



NMR data and X-ray structure determinations show that effective delocalization of the electron density takes place in the compounds  $(R_2N)_2C=E-P=NAr$  (1:  $E = N$ ; 2:  $E = P$ ). Thus, in the molecule  $(Me_2N)_2C=P-P=NAr$  ( $Ar = 2,4,6\text{-tert-Bu}_3C_6H_2$ ) the P-P distance (213 pm) is noticeably shorter than the normal P-P single bond. A strong elongation is observed for formally double  $P=C$  bond (181 pm) in relation to the standard  $P=C$  double bond length. Unlike the compounds 1 and 2 in the compounds  $(R_2N)_2C=E-P=CTms_2$  (3:  $E = N$ ; 4:  $E = P$ ;  $Tms = Me_3Si$ ) a conjugation is mainly localized within the  $(R_2N)_2C=E-$  fragment. In the molecule 4a the P-P bond length (215.5 pm) is close to a usual P-P single bond length and only a slight elongation is observed for the  $P=C$  double bonds. It is essential that the backbone  $N_2C=P-P=CSi_2$  is non-planar; the dihedral angle between the planes of the double bonds is equal to  $51^\circ$ . Apart from X-ray structure investigations this conclusion is supported by chemical behavior of the 2,3-diphosphabutadienes. In particular, the compounds 4a easily reacts with 2,3-dimethylbutadienes to give [2+4]-cycloaddition product:

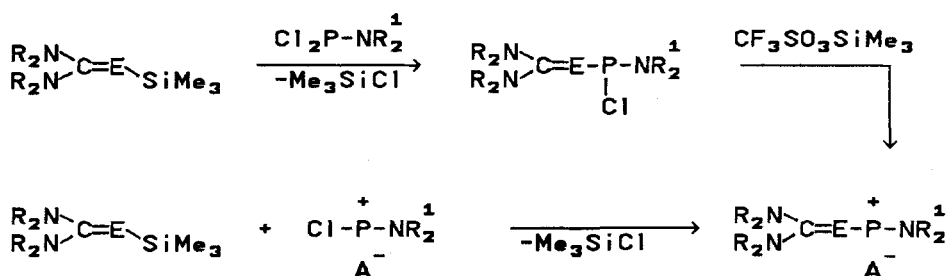


#### Generation and transformations of bis(amino)methyleneamino- and bis(amino)methylenephosphino-substituted phosphonium cations

New reactions including the fast valence isomerization of phosphonium cations to  $\lambda^3$ -iminophosphines and diphosphenes have been discovered:

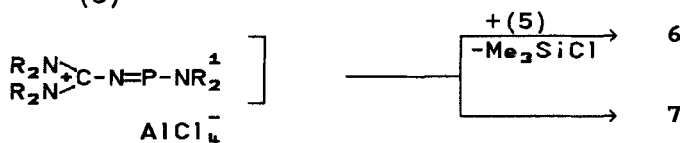
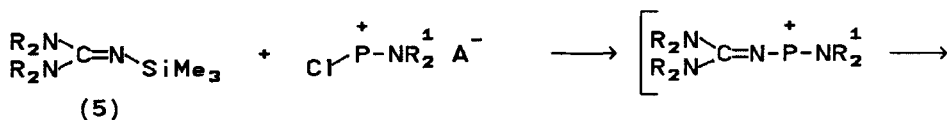


The following pathways for generation of the bis(amino)methyleneamino- and bis(amino)methylenephosphino-substituted phosphonium cations have been employed:

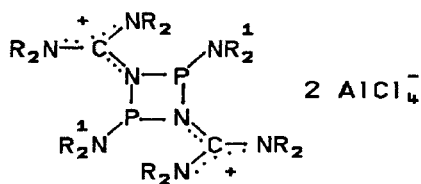


E = N, P; A = AlCl<sub>4</sub><sup>-</sup>, CF<sub>3</sub>SO<sub>3</sub><sup>-</sup>

Depending on the ratio of reactants and the order of their mixing the coupling reactions of (R<sub>2</sub>N)<sub>2</sub>C=NSiMe<sub>3</sub> with Cl-P<sup>+</sup>-NR<sub>2</sub><sup>1</sup> A<sup>-</sup> give various products. If the chloro(amido)phosphenium salt is slowly added into the imine solution, the reaction proceeds in a molar ratio of 1:2 and results in the formation of the cyclic compounds 6. A quite different result is observed in those cases when the imine 5 is added to chloro(amido)phosphenium salt. Addition of 1 equiv of (R<sub>2</sub>N)<sub>2</sub>C=NSiMe<sub>3</sub> at -78°C to Cl-P<sup>+</sup>NR<sub>2</sub><sup>1</sup> AlCl<sub>4</sub><sup>-</sup> leads to the 1,2,3,4-diazadiphosphetidines 7. We have not been able to detect the intermediate species spectroscopically, however, it seems very plausible that the reactions under consideration include the generation of methyleneamino-substituted phosphenium cations and their subsequent valence isomerization into λ<sup>3</sup>-iminophosphines.

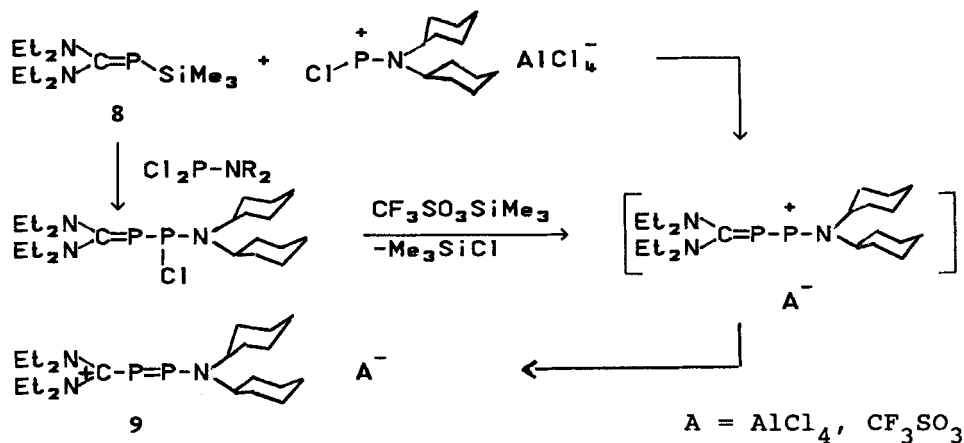


(6)



(7)

Evidence for the transient formation of the compounds  $[(R_2N)_2C-E=P-NR_2]^+ A^-$  was given in the series of diphosphenes. Treatment of the phosphalkene 8 with equimolar amount of  $Cl_2P-NCy_2$  and the following reaction with  $CF_3SO_3SiMe_3$  lead to formation of the diphosphene 9. Alternatively, the methyleneamino-phosphenium cation can be generated by coupling reaction of 8 with  $Cl-P^+-NCy_2 AlCl_4^-$ .



The compound 9 has been clearly characterized by NMR spectroscopy ( $\delta_P$  465 and 210 ppm,  $^1J_{PP}$  525 Hz) and chemical transformations.

#### Acknowledgements

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#### REFERENCES

1. Multiple Bonds and Low Coordination in Phosphorus Chemistry, edited by M. Regitz and O. J. Scherer (Thieme, Stuttgart, 1990), Chap. 4, pp. 157-214; Chap. 8, pp. 293-317.
2. V. D. Romanenko, M. Sanchez, G. Reitel, A. N. Chernega, O.V. Kirichenko, A. V. Ruban, R. Wolf, Heteroatom Chem., in press.
3. V. D. Romanenko, M. Sanchez, T. V. Sarina, M.-R. Mazieres, R. Wolf, Tetrahedron Lett., in press.
4. M. Sanchez, V. Romanenko, M.-R. Mazieres, A. Gudima and L. N. Markovskii, Tetrahedron Lett., **32**, 2775 (1991).
5. A. N. Chernega, A. A. Korkin, N. A. Aksinenko, A.V. Ruban, V.D. Romanenko, Zh. Obshch. Khim., **60**, 2462 (1990)